

MOMENTUM

IB Physics HL

Nuclear physics - Set 1 - Markscheme

E.4-E.5 | First assessment 2025 | Higher Level

Independent examination-style practice material

10 original problems

Task 01 - Balancing a nuclear reaction

Paper 2 style | IB code E.4 | Total [5]

Part	Marking guidance	Marks
(a)	Mass number: $27+4 = 30+A_X$ gives $A_X=1$. Charge: $13+2 = 15+Z_X$ gives $Z_X=0$.	2
(b)	X is a neutron.	1
(c)	Conservation of nucleon number and electric charge.	2

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Task 02 - Energy sharing in alpha decay

Paper 2 style | IB code E.4 | Total [5]

Part	Marking guidance	Marks
(a)	Initial momentum is zero, so the two product momenta are equal and opposite.	1
(b)	For equal p , K is inversely proportional to mass. $K_{\alpha}/K_{\text{Pb}} = 206/4$; gives $K_{\alpha} = 5.304 \text{ MeV}$ and $K_{\text{Pb}} = 0.103 \text{ MeV}$.	4

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Task 03 - The continuous beta spectrum

Paper 2 style | IB code E.4 | Total [6]

Part	Marking guidance	Marks
(a)	C-14 $\rightarrow$ N-14 + electron + electron antineutrino.	2
(b)	The fixed decay energy is shared variably between electron, antineutrino and nuclear recoil.	3
(c)	Atomic number increases by one; mass number is unchanged.	1

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Task 04 - Activity of iodine-131

Paper 2 style | IB code E.4 | Total [6]

Part	Marking guidance	Marks
(a)	$N = (1.00 \times 10^{-9} / 131) N_A = 4.60 \times 10^{12}$.	2
(b)	$\lambda = \ln 2 / (8.02 \times 10^4) = 1.00 \times 10^{-6} \text{ s}^{-1}$.	2
(c)	$A = \lambda N = 4.60 \times 10^6 \text{ Bq}$.	2

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Task 05 - Radiocarbon dating

Paper 2 style | IB code E.4 | Total [5]

Part	Marking guidance	Marks
(a)	$0.250 = (1/2)^2$, so age = $2(5730) = 11460$ years.	2
(b)	Uses $t = -T_{\text{half}} \ln(f)/\ln 2$ for $f=0.25$ and 0.27 , giving approximately 10.8 to 11.5 thousand years.	3

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Task 06 - Mass defect of helium-4

Paper 2 style | IB code E.4 | Total [5]

Part	Marking guidance	Marks
(a)	$\Delta m = 2m_p + 2m_n - m_{\text{nucleus}} = 0.030376 \text{ u.}$	2
(b)	Binding energy = 28.30 MeV and per nucleon = 7.07 MeV.	3

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Task 07 - Interpreting the binding-energy curve

Paper 2 style | IB code E.4 | Total [5]

Part	Marking guidance	Marks
(a)	The maximum is near iron/nickel, A about 56 to 62.	1
(b)	Fusion products have greater binding energy per nucleon, so total mass decreases and energy is released.	2
(c)	Medium-mass fission products are more tightly bound per nucleon than the original heavy nucleus.	2

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Task 08 - Energy from deuterium-tritium fusion

Paper 2 style | IB code E.5 | Total [5]

Part	Marking guidance	Marks
(a)	$\Delta m = 0.018883 \text{ u}$.	2
(b)	$Q = \Delta m c^2 = 17.59 \text{ MeV}$.	2
(c)	A sufficiently high temperature and confinement time/density are required to overcome Coulomb repulsion.	1

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Task 09 - Energy from uranium-235 fission

Paper 2 style | IB code E.5 | Total [6]

Part	Marking guidance	Marks
(a)	$N = N_A/235 = 2.56 \times 10^{21}$.	2
(b)	$E = N(200 \text{ MeV}) = 8.21 \times 10^{10} \text{ J}$.	3
(c)	$E = 22.8 \text{ MWh}$.	1

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Task 10 - Criticality in a chain reaction

Paper 2 style | IB code E.5 | Total [7]

Part	Marking guidance	Marks
(a)	$N_{50}/N_0 = 1.01^{50} = 1.64$.	2
(b)	$N_{50}/N_0 = 0.98^{50} = 0.364$.	2
(c)	$k > 1$ is supercritical, $k < 1$ is subcritical, and steady critical operation requires $k = 1$.	3

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